

## AROMATIC MONOTERPENES FROM *LAVANDULA GIBSONII*\*

S. A. PATWARDHAN and A. S. GUPTA

National Chemical Laboratory, Pune 411008, India

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**Key Word Index**—*Lavandula gibsonii*; Labiatae; aromatic monoterpenes; 3-hydroxy- $\alpha,\alpha,4$ -trimethyl benzyl alcohol; 3-hydroxy- $\alpha,\alpha,4$ -trimethyl benzyl methyl ether; 3-hydroxy- $\alpha,4$ -dimethyl styrene.

**Abstract**—Three aromatic monoterpenes, not reported previously as natural products, together with ursolic acid, were isolated from *Lavandula gibsonii*. They were characterized as 3-hydroxy- $\alpha,\alpha,4$ -trimethyl benzyl alcohol, 3-hydroxy- $\alpha,\alpha,4$ -trimethyl benzyl methyl ether and 3-hydroxy- $\alpha,4$ -dimethyl styrene.

In continuation of our work on *L. gibsonii* [1] we report here the isolation and characterization of a new aromatic monoterpene, 3-hydroxy- $\alpha,\alpha,4$ -trimethyl benzyl methyl ether (**2**) along with two other similar compounds, 3-hydroxy- $\alpha,\alpha,4$ -trimethyl benzyl alcohol (**1**) and 3-hydroxy- $\alpha,4$ -dimethyl styrene (**3**). These compounds do not appear to have been reported previously as natural products but have been synthesized [2]. Column chromatography (Si gel) of the crude acetone extract of *L. gibsonii* resulted in isolation of compounds **1**–**3**. Compound **1**, mp 105–108°, analysed for  $C_{10}H_{14}O_2$  ( $M^+$  at  $m/z$  166). The IR spectrum showed the presence of only a hydroxyl group ( $3226\text{ cm}^{-1}$ ) and no other functional group containing oxygen. Bands at 1613, 1515 and  $1451\text{ cm}^{-1}$  in the IR spectrum and the  $^1\text{H}$  NMR spectral pattern indicated the aromatic nature of the compound. A 1,3,4-substitution pattern of an aromatic ring was indicated by the following signals in the  $^1\text{H}$  NMR spectrum at  $\delta$ 6.65 (1H, *dd*,  $J = 8, 2\text{ Hz}$ ), 6.91 (1H, *d*,  $J = 7\text{ Hz}$ ) and 6.93 (1H, *d*,  $J = 2\text{ Hz}$ ). Three methyls were indicated by a 3H singlet at 2.16 and a 6H singlet at 1.53. Acetylation of **1** yielded three products. A hydroxy aromatic monoacetate (**6**) [ $\delta$ 2.03 (Ar–OCOMe),  $3571\text{ cm}^{-1}$  (OH)], a diacetate (**5**) [ $\delta$ 1.90 (–OCOMe) and 2.18 (Ar–OCOMe)] and the third product characterized as **4** [ $\delta$ 2.16 ( $2 \times \text{Me}$ ), 2.25 (Ar–OCOMe) and 4.90, 5.18 ( $=\text{CH}_2$ )]. Considering all these data, the two structures **1** and **1a**, analogous to carvacrol and thymol, respectively, seem to be biogenetically possible. Structure **1**, 3-hydroxy- $\alpha,\alpha,4$ -trimethyl benzyl alcohol, was confirmed by its synthesis from 3-hydroxy-4-methyl-acetophenone [3] by alkylation with methyl magnesium iodide.

Compound **2**, mp 89–92°,  $C_{11}H_{16}O_2$  ( $M^+$  at  $m/z$  180), showed an additional methyl signal at  $\delta$ 3.08 (–OMe) in its  $^1\text{H}$  NMR spectrum by comparison with that of **1**, along with a hydroxyl band ( $3400\text{ cm}^{-1}$ ) in the IR spectrum. Acetylation of **2** gave an acetate **9** ( $\delta$ 2.30) and methylation yielded an ether (**8**) which exhibited an additional –OMe ( $\delta$ 3.51). Compound **1** also gave the diether **8** in addition to a monoether **7**. This data clearly indicates that the isopropanolic hydroxyl of **1** is selectively methylated in

nature to produce **2**. Thus, **2** was identified as 3-hydroxy- $\alpha,\alpha,4$ -trimethyl benzyl methyl ether.

Compound **3**, a pale yellow viscous liquid, bp 120–125°/0.6 mm,  $C_{10}H_{12}O$  ( $M^+$  at  $m/z$  148), possessed an aromatic methyl ( $\delta$ 2.00), one methyl at  $\delta$ 1.88, a methylene group ( $\delta$ 4.46, 4.73) and a hydroxyl ( $3448\text{ cm}^{-1}$ ). Compound **4** on hydrolysis produced **3** and, hence, **3** was characterized as 3-hydroxy- $\alpha,4$ -dimethyl styrene.

In our first attempt to methylate **1** with dimethyl sulphate, instead of a mono- or diether, a solid, mp 104–105° was obtained which was identified as 4-methoxy-1,1,3,5-tetramethyl-3-(3-methoxy-4-methyl phenyl) indane (**10**) [4]. It has been observed that **1**, on keeping in deuteriochloroform solution for a long time, dimerized to give the corresponding hydroxyl compound **11**, mp 170–171°.

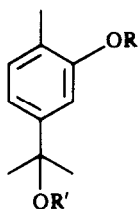
### EXPERIMENTAL

General exptal details have been described previously [1].

Earlier fractions ( $C_6H_6$ ), in the chromatography of fraction B<sub>3</sub> (35 g) [1] eluted methyl-3-acetoxy-16-methyl-heptadecanoate. A white solid (2 g), obtained from the later fractions ( $C_6H_6$ ), was acetylated ( $\text{Ac}_2\text{O}$  and pyridine), esterified ( $\text{CH}_2\text{N}_2$ ) and characterized as acetyl methyl ursolate mp 235–238° ( $C_6H_6$ –petrol) (lit. 239–241° [5]).

$C_6H_6$ – $\text{Me}_2\text{CO}$  (98:2 and 96:4) gave a viscous dark material (10 g) which showed three spots in TLC ( $C_6H_6$ –EtOAc, 75:25). These were separated by passing through a Si gel column.  $C_6H_6$ – $\text{Me}_2\text{CO}$  (98:2) eluted **3** [0.550 g, bp 120–125° (bath-temp.)/0.6 mm] [lit. [2] 60–100° (bath)/0.01 mm];  $^1\text{H}$  NMR:  $\delta$ 1.88 (3H, s), 2.00 (3H, s), 4.46, 4.73 (2H), 6.10 (1H), 6.25 (2H). MS  $m/z$  (rel. int.): 148 [ $M^+$ ] (100), 133 (75), 131 (44), 115 (44), 108 (66), 107 (55), 105 (58), 91 (32), 77 (41), 51 (44).  $C_6H_6$ – $\text{Me}_2\text{CO}$  (94:6) gave a white solid **2** (2 g) mp 89–92° (petrol). (Found: C, 72.91; H, 8.88. Calc. for  $C_{11}H_{16}O_2$ : C, 73.30; H, 8.95%). IR  $\nu_{\text{max}}^{\text{Nujol}}\text{ cm}^{-1}$ : 3400, 2994, 1590, 1449, 1408, 1267, 1117, 1058, 893, 805.  $^1\text{H}$  NMR:  $\delta$ 1.56 (6H, s), 2.18 (3H, s), 3.08 (3H, s), 6.53 (1H, *dd*,  $J = 7, 2\text{ Hz}$ ), 6.88 (1H, *d*,  $J = 7\text{ Hz}$ ) 7.05 (1H, *d*,  $J = 2\text{ Hz}$ ). MS  $m/z$  (rel. int.): 180 [ $M^+$ ] (98), 165 (100), 150 (41), 149 (91), 135 (82), 133 (25), 121 (52), 107 (23), 91 (38), 77 (54.5). Acetate **9**, ( $\text{Ac}_2\text{O}$  pyridine); bp 130–135°/1.0 mm;  $^1\text{H}$  NMR:  $\delta$ 1.53 (6H, s), 2.16 (3H, s), 2.30 (3H, s), 3.03 (3H, s), 6.93 (1H), 7.10 (2H).

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**1** R = R' = H

**2** R = H, R' = Me

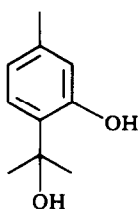
**5** R = R' = Ac

**6** R = Ac, R' = H

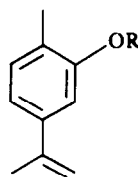
**7** R = Me, R' = H

**8** R = R' = Me

**9** R = Ac, R' = Me

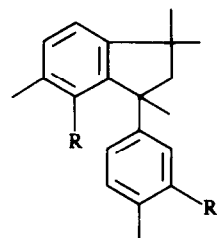


**1a**



**3** R = H

**4** R = Ac



**10** R = R' = OMe

**11** R = R' = OH

$\text{C}_6\text{H}_6\text{-Me}_2\text{CO}$  (90:10) yielded a white solid **1** (4 g), mp  $105\text{--}108^\circ$  (lit. [2]  $105\text{--}106^\circ$ ) UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 207 (4.56), 221 (4.49), 277 (3.97).  $^1\text{H}$  NMR:  $\delta$  1.53 (6H, s), 2.16 (3H, s), 6.65 (1H, dd,  $J = 8, 2$  Hz), 6.91 (1H, d,  $J = 7$  Hz), 6.97 (1H, d,  $J = 2$  Hz). MS  $m/z$  (rel. int.): 166  $[\text{M}]^+$  (66), 151 (100), 149 (20), 148 (40), 135 (20), 133 (26), 108 (33), 107 (15), 91 (26), 77 (40). Acetylation of **1** (0.2 g), with  $\text{Ac}_2\text{O}$ -pyridine gave **4** (0.02 g); bp  $120\text{--}125^\circ/1.0$  mm;  $^1\text{H}$  NMR:  $\delta$  2.16 (6H, s), 2.25 (3H, s), 4.90, 5.18 (2H), 6.85 (1H), 6.95 (2H). Compound **5** (0.09 g), bp  $145\text{--}150^\circ/0.9$  mm.  $^1\text{H}$  NMR:  $\delta$  1.71 (6H, s), 1.90 (3H, s), 2.08 (3H, s), 2.18 (3H, s), 6.73 (1H), 6.90 (2H). Compound **6** (0.03 g), bp  $135\text{--}140^\circ/0.8$  mm.  $^1\text{H}$  NMR:  $\delta$  1.43 (6H, s), 2.03 (3H, s), 2.16 (3H, s), 6.83 (1H), 6.90 (2H). Etherification (NaH, DMSO, MeI) of 0.2 g of **1** gave **8** (0.082 g), bp  $90\text{--}95^\circ/1.0$  mm,  $^1\text{H}$  NMR:  $\delta$  1.5 (6H, s), 2.20 (3H, s), 3.03 (3H, s), 3.85 (3H, s), 6.75 (1H, dd,  $J = 8, 2$  Hz), 6.85 (1H, d,  $J = 2$  Hz), 6.98 (1H, d,  $J = 8$  Hz) and **7** (0.1 g), mp  $62^\circ$  (petrol) [6] Methylation ( $\text{Me}_2\text{SO}_4$ ,  $\text{Me}_2\text{CO}$ ,  $\text{K}_2\text{CO}_3$ ) of **1** (0.01 g) yielded **10**, mp  $104\text{--}105^\circ$  (lit. [4] mp  $95\text{--}96^\circ$ ).  $^1\text{H}$  NMR:  $\delta$  1.01 (3H, s), 1.26 (3H, s), 1.56 (3H, s), 1.98, 2.03 (2H,  $2ds$ ,  $J = 12$  Hz), 2.05 (3H, s), 2.1 (3H, s), 3.58 (3H, s), 3.75 (3H, s), 6.26 (2H, s), 6.28–6.66 (3H). Compound **11** was isolated from a  $\text{CDCl}_3$  soln of **1** by

chromatography over Si gel as a white crystalline solid, mp  $170\text{--}171^\circ$  (petrol- $\text{C}_6\text{H}_6$ )  $^1\text{H}$  NMR:  $\delta$  1.0 (3H, s), 1.25 (3H, s), 1.56 (3H, s), 1.98, 2.31 (2H,  $2ds$ ,  $J = 10$  Hz), 2.16 (3H, s), 2.21 (3H, s), 4.43 (2H, br, exchanges with  $\text{D}_2\text{O}$ ) 6.4 (2H, s), 6.53–6.83 (3H, m). MS  $m/z$  (rel. int.): 296  $[\text{M}]^+$  (48), 281 (100), 266 (9.5), 265 (9.5), 251 (11.5), 189 (22.5), 173 (66.5), 157 (27), 133 (14), 128 (30).

**Synthesis of 1.** 3-Hydroxy-4-methyl-acetophenone (1.5 g, THF 20 ml) with MeMgI (from 2 g Mg and 9.7 g MeI in  $\text{Et}_2\text{O}$ ) yielded, after work-up with aq.  $\text{NH}_4\text{Cl}$ , **1** (1.29 g), mp, mmp  $103\text{--}105^\circ$  (Petrol- $\text{C}_6\text{H}_6$ ).

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